

Laboratory Instructions:

→ grads:

Homeworks	10 %
Quizzes	10 %
Reports	40 %
Final exam	40 %

→ Reports:

* Cover page: Name, number, date
Partner name, experiment name

* Abstract: a brief summary explaining

- The aim of the experiment.
- The method used
- The main results

* Theory

* Procedure

* Data

* Calculations

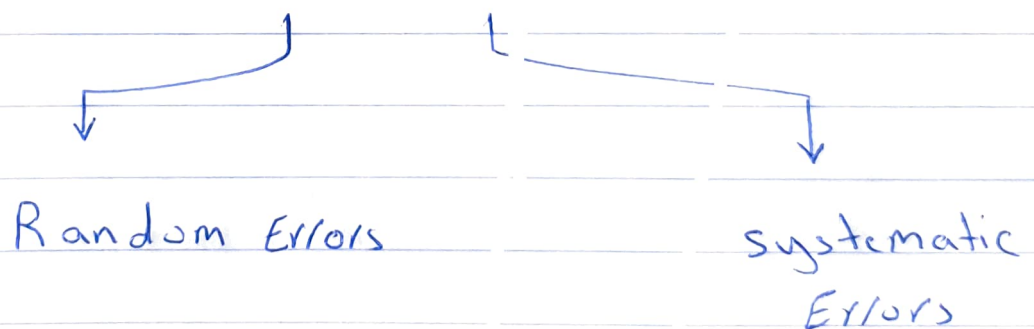
* Conclusion.

Measurements & Uncertainties

* Sources of errors :

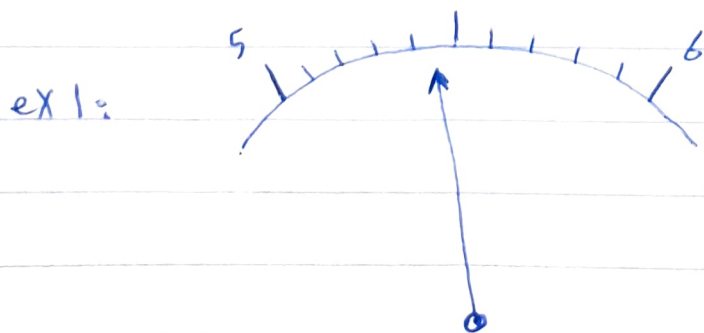
- choice of instruments
- Environment
- The way the experiment is done
- Experimenter

* types of errors



4.1 : Random Errors .

4.1.1 Uncertainty in a measurement

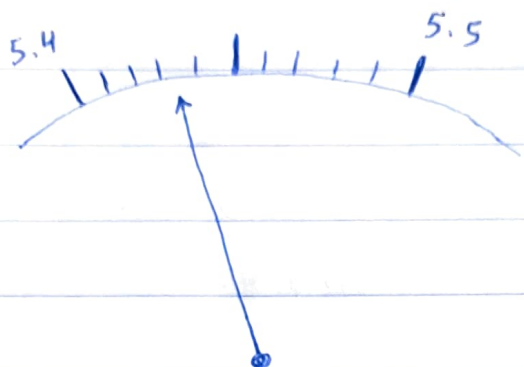


5.44 or 5.45
or 5.43

There is uncertainty about the reading of the vol.

by about ± 0.1 Volt. $\Rightarrow V = 5.4 \pm 0.1$ Volt

ex 2:



5.432 or 5.433
or 5.434

I'm uncertain about the last figure

There is uncertainty by about ± 0.01

$\Rightarrow V = 5.43 \pm 0.01$ Volt

The estimated uncertainty depends on how finely the scale is divided

* If only one measurement of a quantity is made?
The only way to find the uncertainty is to estimate it.

* now, if several measurements of the same quantity are made: x_1, x_2, \dots, x_n

\Rightarrow The best estimate of the true value is the average

value, defined as: $\bar{x} = \frac{1}{N} \sum_{i=1}^N x_i$

$$= \frac{x_1 + x_2 + x_3 + \dots + x_N}{N}$$

• The best estimate of the uncertainty is the sample standard deviation:

$$\Delta_s = \sqrt{\frac{1}{N-1} \sum_{i=1}^N (x_i - \bar{x})^2}$$

It means that the probability is about 2/3 (66.7%) that any one measurement does not differ from another by more than Δ_s .

4.1.2 Uncertainty in the mean = "standard deviation of the mean"

$$\Delta_m = \frac{\Delta_s}{\sqrt{N}}$$

• it shows the degree of confidence that the average value is close to the true value.

• clearly the more measurements made, the closer the mean to the true value.

ex: suppose five students measured the voltage across a resistance (R)

student No 1 2 3 4 5

Voltage (Volts) 5.45 5.44 5.43 5.46 5.43

st. #	volt	$x_i - \bar{x}$	$(x_i - \bar{x})^2$
1	5.45	5.45 - 5.442	$(0.008)^2$
2	5.44	5.44 - 5.442	$(-0.002)^2$
3	5.43	⋮	⋮
4	5.46	⋮	⋮
5	5.43	⋮	⋮
$N = 5$	$\bar{x} = \frac{1}{N} \sum_{i=1}^N x_i$		$\sum_{i=1}^5 (x_i - 5.442)^2 = \square$
	$\bar{x} = \frac{1}{5} \sum_{i=1}^5 x_i$		

$$\bar{x} = 5.442 \text{ V}$$

$$s_s = \sqrt{\frac{1}{(5-1)} \sum_{i=1}^5 (x_i - 5.442)^2}$$

$$s_s = 0.013 \text{ volt}$$

$$s_m = \frac{s_s}{\sqrt{N}} = \frac{0.013}{\sqrt{5}} = 0.0058 \text{ volt}$$

* Back to Appendix C [P 87-88] for using calculator.

4.1.3 Importance of knowing the uncertainty:

Consider the measurements of the length of a rod at two different temperatures:

length (cm)	98.025	98.034
Temperature ($^{\circ}$ C)	10.0	20.0

Does the length of the rod depend on the temp.?

if the uncertainty is ± 0.01 cm
 \rightarrow we can't answer the question

since the difference between the two measurements is smaller than their uncertainties.

i.e.: if we take the length 98.025 cm
then the true value is 98.025 ± 0.01 cm

$$\downarrow$$
$$[98.025 - 0.01, 98.025 + 0.01]$$

The true value lay between $[98.015, 98.035]$

4.2 Systematic Errors:

Random errors

→ are always present in an experiment

→ They are equally likely to be positive or negative & cause several measurements to spread out around the true value

Systematic errors

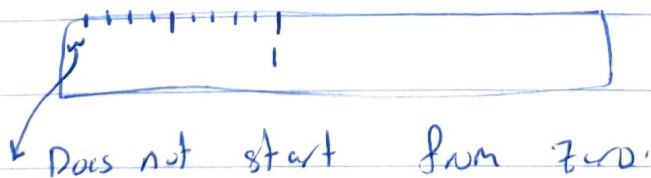
it depends on the experimenter & environment.

shift the measured values to be bigger or smaller of the true value

→ examples:

1. uncalibrated tools

2. worn off ruler used to measure the length of a rod.



3. the way the quantity is measured.

H.W [Solu Ex1, Ex2 p [20, 22]]

4.2.1 Precision and Accuracy.

↓
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* Small Random Errors means High Precision.

ex: two students made two measurements of the voltage.

$$V_1 = 5.443 \pm 0.002 \text{ Volt}$$

$$V_2 = 5.44 \pm 0.03 \text{ Volt}$$

st. 1 has less random error compared with st. 2
 V_1 more precise than V_2 .

* Negligible systematic errors means High Accuracy

ie: if the measured value is close to the true value \Rightarrow the measurement is said to be accurate

\Rightarrow precise but not accurate

ie. $D = 2.13 \pm 0.01 \text{ cm}$ [the true value is 2.15 cm]

Due to systematic errors.

accurate but not precise

ie. $g = 9.9 \pm 0.2 \text{ m/s}^2$, The true value is 9.82 m/s^2

Due to the big uncertainty, but the measurement is accepted. since $g \in [9.9 - 0.2, 9.9 + 0.2]$
 $g \in [9.7, 9.11]$

4.2.2 A comparison between measured & accepted values !!

* Discrepancy test : $|\text{measured value} - \text{accepted value}| \leq 2 \text{ uncertainty}$

ex: Two students A & B measured the speed of sound, $V_A = 338 \pm 2 \text{ m/s}$
 $V_B = 325 \pm 5 \text{ m/s}$

The true value $V = 331 \text{ m/s}$

which measurement is accepted ??

By using Discrepancy test.

$$D_A = |338 - 331| \stackrel{??}{\leq} 2(2) \quad \rightarrow \quad D V_A$$

~~7~~ ~~4~~ \rightarrow not accepted.

$$D_B = |331 - 325| \stackrel{??}{\leq} 2(5)$$

$$6 \leq 10 \quad \text{its accepted } \checkmark$$

* Or in other words, if the result = $\bar{x} \pm \Delta_m$
 $\Delta_m = \Delta x = \text{uncertainty}$

The result is accepted if

$$\bar{x} - 2\Delta_m \leq \text{True value} \leq \bar{x} + 2\Delta_m$$

$$-\bar{x} \quad -\bar{x} \quad -\bar{x}$$

$$-2\Delta_m \leq \text{True value} - \bar{x} \leq 2\Delta_m$$

$$\Rightarrow |\text{True value} - \bar{x}| \leq 2\Delta_m$$

4.3 Significant figures:

if I have $\Delta L = \pm 0.1 \text{ cm}$, $L = 2.435 \text{ cm}$
 \downarrow meaningless

Because 4 is uncertain by ± 0.1

$\Rightarrow 2, 4$ are significant figures

$$\Rightarrow L = 2.4 \pm 0.1 \text{ cm}$$

if I use micrometer $\Delta L = \pm 0.01 \text{ mm}$
 $= \pm 0.001 \text{ cm}$

\Rightarrow 2, 4, 3, 5 are significant figures.

$L = 2.435 \pm 0.001 \text{ cm}$ [4 significant fig]
[I'm uncertain about "5"]

ex: 4.7 ± 0.3 (2 sig. fig.)

473 ± 2 (3 sig. fig.)

472.84 ± 0.03 (5 sig. fig.)

0.47 ± 0.03 (2 sig. fig.)

$472 \pm 3 = (4.72 \pm 0.03) \times 10^2$ (3 sig. fig.)

* Wrong way to write sig. fig.

$\rightarrow 4.7 \pm 0.03$ X $\Rightarrow 4.70 \pm 0.03$
[3 sig. fig.]

$\rightarrow 472.8 \pm 0.04$ X $\Rightarrow 472.80 \pm 0.04$
[5 sig. fig.]

$\rightarrow 0.47 \pm 0.3$ X $\Rightarrow 0.5 \pm 0.3$
[1 sig. fig.]

$\rightarrow 473 \pm 30$ X $\rightarrow (47.3 \pm 3) \times 10$
↓
meaningless
 $(47 \pm 3) \times 10$ (2 sig. fig.)
 $= 470 \pm 30$

4.3.1 Significant figures in calculated values:

4.3.1.1 Rounding:

ex: $X = 4.374 \pm 0.1$

we must round the last significant fig "3" !!

if the first non significant fig "7" > 5 , we round "3"
 $\Rightarrow X = 4.4 \pm 0.1$

if the first non significant fig < 5 , we fix the last sig. fig.

ex: $L = 4.351 \pm 0.03$
the last sig. fig < 5

$\Rightarrow L = 4.35 \pm 0.03$

ex: $3.7 \pm 1 \Rightarrow 4 \pm 1$
last sig. fig $7 > 5$

$51.36 \pm 0.2 \Rightarrow 51.4 \pm 0.2$
last sig. fig $6 > 5$

$33.1 \pm 10 \Rightarrow (3.31 \pm 1) \times 10$
last sig. fig, $3 < 5$

4.3.1.3 Multiplication & division

* The number of significant fig. in the result will be given by the fewest number of sig. fig.

$$\text{ex: } R = \frac{1.8 \times 2.346}{7.86} = 0.537$$

but fewest number of sig. fig [1.8] = 2

⇒ the result is rounded to only two sig. fig.

$$R \approx 0.54$$

$$\text{ex: } \sin(12^\circ) = 0.21 \quad (\theta \text{ has 2 sig. fig.} \\ \Rightarrow \text{the result must have 2 sig. fig.})$$

the same for Cos

$$\text{ex } \sqrt{\frac{2.3 \times 4.57}{1.2}} = 3.0 \quad (\text{must have 2 sig. fig.})$$

4.3.2 Significant figures in the uncertainty

* The uncertainty in a result should be rounded to one significant fig.

$$\text{ex: } \Delta X = \pm 0.0237 \Rightarrow \Delta X = \pm 0.02$$

$$\Delta L = \pm 0.0412 \Rightarrow \Delta L = \pm 0.04$$

* An exception to this rule if the leading digit in the uncertainty is "1", then we keep two sig. fig.

$$\text{ex: } \Delta y = \pm 0.147 \Rightarrow \Delta y = \pm 0.15$$

4.4 Combining Uncertainties:

4.4.1 Addition & subtraction:

$$\text{if } R = X \pm Y \Rightarrow \Delta R = \Delta X \pm \Delta Y$$

$$\text{ex: } X = 3.4 \pm 0.2$$

$$Y = 7.5 \pm 0.3$$

$$R = X - Y = 3.4 - 7.5 = -4.1$$

$$\Delta R = \Delta X + \Delta Y = 0.2 + 0.3 = 0.5$$

4.4.2 Constant multipliers

if $R = ax + by$; a, b constants

$$DR = a DX + b DY$$

ex: $X = 57 \pm 2$

$$Y = 23 \pm 3$$

$$R = X - 2Y = 57 - 2 \times 23 = 11$$

$$DR = DX + 2DY = 2 + 2 \times 3 = 8$$

$$R = 11 \pm 8$$

4.4.3 Multiplication & division

if $A = xy$ we find DA by direct differentiation with respect to x & y

$$DA = \frac{\partial A}{\partial x} DX + \frac{\partial A}{\partial y} DY$$

$$DA = Y DX + X DY$$

dividing by $A \Rightarrow \frac{DA}{A} = \frac{Y}{xy} DX + \frac{X}{xy} DY$

$$\Rightarrow \frac{DA}{A} = \frac{DX}{x} + \frac{DY}{y}$$

$$\text{ex: } L = 8.27 \pm 0.05, \quad W = 5.12 \pm 0.02$$

$$A = LW = 8.27 \times 5.12 = 42.3424$$

$$\frac{\Delta A}{A} = \frac{\Delta L}{L} + \frac{\Delta W}{W}$$

$$\frac{\Delta A}{42.3424} = \frac{0.05}{8.27} + \frac{0.02}{5.12}$$

$$\Rightarrow \Delta A = 0.4214 \Rightarrow \Delta A = 0.4 \quad (1 \text{ sig. fig})$$

$$A = 42.3 \pm 0.4$$

4.4.4 Raising to a power

$$R = x^n y^m z^l \quad \text{by differentiation}$$

$$\Delta R = \left| \frac{\partial R}{\partial x} \right| \Delta x + \left| \frac{\partial R}{\partial y} \right| \Delta y + \left| \frac{\partial R}{\partial z} \right| \Delta z$$

$$\Delta R = |n x^{n-1} y^m z^l| \Delta x + |x^n m y^{m-1} z^l| \Delta y + |x^n y^m l z^{l-1}| \Delta z$$

dividing by R

$$\frac{DR}{R} = \left| \frac{n}{x} \right| dx + \left| \frac{m}{y} \right| dy + \left| \frac{l}{z} \right| dz$$

ex: $R = x^2$

$$\Rightarrow \frac{DR}{R} = 2 \frac{dx}{x}$$

ex: $R = z^2 y^3 / x^4$

$$\frac{DR}{R} = 2 \frac{dz}{z} + 3 \frac{dy}{y} + 4 \frac{dx}{x}$$

4.4.5 Other functions:

①

$$R = \sin(x) \Rightarrow dR = |\cos(x)| dx$$

$$R = \cos(x) \Rightarrow dR = |\sin(x)| dx$$

$$R = \tan(x) \Rightarrow dR = |\sec^2(x)| dx$$

$$R = \cot(x) \Rightarrow dR = |\csc^2(x)| dx$$

$$R = \sec(x) \Rightarrow dR = |\sec(x) \tan(x)| dx$$

* x must be in Radians

if x in degrees \Rightarrow multiply by $\frac{\pi}{180}$

$$\text{ex: } \theta = 80^\circ \pm 1^\circ$$

$$R = \sin \theta = \sin(80^\circ) = 0.984810$$

$$dR = \cos \theta \left(d\theta \frac{\pi}{180} \right) = \cos(80^\circ) \left(\frac{\pi}{180} \times 1 \right)$$

$$= 0.0030307$$

$$dR = 0.003$$

$$R = 0.985 \pm 0.003$$

2. Natural logarithm

$$R = \ln(x) \Rightarrow dR = \frac{dx}{x}$$

3. exponential function

$$\text{if } R = e^x \Rightarrow dR = e^x dx$$

H. w

Solve: Ex 3 P 32

Ex 4 P 36